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| **Name the Compounds** | **Write the Formula for the Compounds** |
| FeO Iron (II) Oxide | Chromium (III) chlorideCrCl3 |
| SnS2 Tin (IV) sulfide | Lead (II) SulfidePbS |
| Cr2S3Chromium (III) Sulfide | Copper (II) fluorideCuF2 |
| SnF2Tin (II) fluoride | Lead (IV) ChloridePbCl4 |
| CuOCopper (II) oxide | Sodium sulphideNa2S |
| CuClCopper (I) chloride | Chromium (II) oxideCrO |
| MnO2Manganese (IV) oxide | Iron (II) OxideFeO |
| PbBr2Lead (II) bromide | Gold (I) phosphide Au3P |
| Fe2O3Iron (III) oxide | Mercury (II) nitrideHg3N2 |
| AgISilver iodide | Vanadium (V) fluorideVF5 |
| ZnF2Zinc fluoride | Tin (IV) iodideSnI4 |
| ZrO2 Zirconium oxide | Chromium (VI) oxideCrO3 |
| TiBr4Titanium (IV) bromide | Iron (III) nitrideFeN |
| Pd3N2Palladium (II) nitride | Zirconium (IV) bromideZrBr4 |
| NiCl2Nickel (II) chloride | Cobalt (III) phosphideCoP |
| AuPGold (III) phosphide | Arsenic (III) chlorideAsCl3 |
| FeF3Iron (III) fluoride | Platinum (IV) sulfidePtS2 |
| CrNChromium (III) nitride | Iron (II) sulfideFeS |