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| **Name the Compounds** | **Write the Formula for the Compounds** |
| FeO  Iron (II) Oxide | Chromium (III) chloride  CrCl3 |
| SnS2  Tin (IV) sulfide | Lead (II) Sulfide  PbS |
| Cr2S3  Chromium (III) Sulfide | Copper (II) fluoride  CuF2 |
| SnF2  Tin (II) fluoride | Lead (IV) Chloride  PbCl4 |
| CuO  Copper (II) oxide | Sodium sulphide  Na2S |
| CuCl  Copper (I) chloride | Chromium (II) oxide  CrO |
| MnO2  Manganese (IV) oxide | Iron (II) Oxide  FeO |
| PbBr2  Lead (II) bromide | Gold (I) phosphide  Au3P |
| Fe2O3  Iron (III) oxide | Mercury (II) nitride  Hg3N2 |
| AgI  Silver iodide | Vanadium (V) fluoride  VF5 |
| ZnF2  Zinc fluoride | Tin (IV) iodide  SnI4 |
| ZrO2  Zirconium oxide | Chromium (VI) oxide  CrO3 |
| TiBr4  Titanium (IV) bromide | Iron (III) nitride  FeN |
| Pd3N2  Palladium (II) nitride | Zirconium (IV) bromide  ZrBr4 |
| NiCl2  Nickel (II) chloride | Cobalt (III) phosphide  CoP |
| AuP  Gold (III) phosphide | Arsenic (III) chloride  AsCl3 |
| FeF3  Iron (III) fluoride | Platinum (IV) sulfide  PtS2 |
| CrN  Chromium (III) nitride | Iron (II) sulfide  FeS |